The allylamine class of antifungals is highly lipophilic. They inhibit fungal growth by disrupting sterol biosynthesis. It abrogates the formation of ergosterol by inhibiting squalene epoxidase, the catalytic enzyme responsible for converting squalene to 2, 3-oxidosqualene (an ergosterol precursor). These drugs are indicated to treat fungal skin and nail infections caused by Trichophyton species, Microsporum canis, Epidermophyton floccosum, and Tinea species. They also treat yeast infections of the skin caused by Candida species (1).

Recently, there has been an increased interest in drug administration via the skin (topical delivery) for both local therapeutic effects and for systemic delivery (transdermal delivery). We can measure the permeation of chemicals through the skin by in vivo and in vitro techniques. Frequently, this has been done by in vitro techniques because of the simplicity of the experimental conditions. The in vitro release test (IVRT) is used to monitor the release and diffusion of drug products from semisolid dosage forms and has long been considered as a valuable tool in formulation development. IVRT has also been used to screen formulations to select promising candidates and, importantly, is accepted with the purpose of obtaining a waiver of bioequivalence studies following post-approval changes (SUPAC-SS) guidance. The obtained results confirm the competence of the IVRT method for the assessment of product sameness.

ABSTRACT
Excipients play a very important role in the release pattern of an active pharmaceutical ingredient from topical semisolid dosage forms, and their physical and chemical properties can influence the release. The aim of this paper was to provide a validated, sensitive, and reproducible method to assess the in vitro release rate of an antifungal drug (Terbinafine) from controlled drug delivery systems (cutaneous and film-forming solution) and to prove product sameness. The samples obtained from in vitro testing were analyzed through a high performance liquid chromatographic (HPLC) system coupled with UV spectrometer at a wavelength of 283 nm. A Franz diffusion cell (FDC) system was used for the dissolution test. We recorded the drug release from the formulation for 6 hours. The release rate obtained from cutaneous and film-forming solutions were compared statistically to depict the sameness. The results met the relevant acceptance criteria (i.e., 90% confidence interval, falling within the limits of 75–133%) as defined in the scale-up and post-approval changes (SUPAC-SS) guidance. The obtained results confirm the competence of the IVRT method for the assessment of product sameness.

KEYWORDS: In vitro release testing, Franz diffusion cell, apparatus qualification, terbinafine, dissolution
or manufacturing process/site. Besides its use in quality control, IVRT can also optimize formulation during the early stages of development. IVRT is a cost-effective alternative for providing some predictive estimates regarding the in vivo performance of a drug product.

The development of controlled drug delivery systems has generated considerable interest in pharmaceutical science in recent years. In particular, the transdermal drug delivery has attracted researchers with multiple approaches because multiple dosing or insufficient drug delivery often results in low therapeutic effects. Among these techniques, films (patches) and gels have been extensively designed for skin diseases or wound care in the past decades. These dosage forms can also contain drugs for therapeutic applications. Fortunately, the combined advantages of both film and hydrogels were found in film-forming solutions (FFSs) for transforming the drug via thin film. In fact, FFSs contain three key components, i.e., the drug, film-forming polymer, and solvent(s). Upon coming in contact with the target site (usually skin), the solvent will evaporate to form a film-loading drug (3).

The FFS technique of drug delivery is an effective and novel approach for the delivery of drug in skin. This technique was developed and categorized in agreement with their mechanical properties and water vapor permeability. For making FFSs, the drug and film-forming excipient are dissolved/dispersed in a volatile solvent(s). The liquid state of the FFS depends on the solubility of drug/excipient or dispersions of encapsulated drug micro particles/nanoparticles in solvents. After coming in contact with the skin, the solvents evaporate and form a film with excipient. With FFSs, the polymer can have a tight contact via molecular interactions to build an even film or smooth film, thus facilitating prolong release (3).

The current research work aims to qualify and validate the IVRT and high performance liquid chromatography (HPLC) method and to specify the procedure for providing the product sameness by comparing the release rate of terbinafine (TBF, 1%) from its cutaneous and film-forming solutions. The selection of receptor medium for estimating the release rate of TBF through synthetic membrane was challenging, as TBF degraded rapidly in phosphate buffered saline/aqueous media. However, in organic/organic-aqueous media, over 30% of cumulative release was observed, showing deviation from Higuchi theory (2, 4). Moreover, the coefficient of determination ($R^2$) was less than 0.90% and inter cell precision greater than 15%. These issues were resolved by lowering the pH of receptor medium to approximately 5.5 using ascorbic acid in an optimized quantity, which prevented the degradation of TBF. Moreover, by lowering the pH of receptor solution, precise and accurate results were obtained, and the sink condition was maintained throughout the experimental duration.

An extensive literature search revealed that no IVRT method has been published for estimating the release rate of TBF from cutaneous and film-forming solutions. We found some analytical methods for determination of TBF via UV spectrophotography or chromatographically (5).

**MATERIALS AND METHODS**

**Chemicals and Reagents**

Terbinafine hydrochloride (working standard) was procured from Sun Pharmaceutical Industries Ltd., Baddi, India. Acetonitrile and methanol (HPLC Grade) were purchased from Sigma Aldrich Chemical Pvt. Ltd., Bengaluru, India. Trifluoroacetic acid and ascorbic acid (analytical grade) were purchased from Thermo Fisher Scientific, Haryana, India. Phosphate buffered saline was purchased from Sisco Research Laboratories Pvt. Ltd., Mumbai, India.

**Equipment**

IVRT experiments were performed in a Franz diffusion cell (FDC) from PermeGear, Pennsylvania, USA. Synthetic membranes for IVRT experiments, i.e., Ultipor N66, Nylon 6,6 (0.2 µm × 25 mm, Lot No. IN15000705; 0.2 µm × 25 mm, Lot No. IN12000094), Tuffryn HT-200, (0.2 µm × 25 mm, Lot No. T30120), and Supor 200 (0.2 µm × 25 mm, Lot No. IN14000045) were purchased from Pall Life Sciences, Mumbai, India. For temperature monitoring of synthetic from Metravi, MT-4, West Bengal, India was used. For quantification of IVRT samples, HPLC system coupled with UV detector from Shimadzu, Mumbai, India was used along with Analyst 1.6.3 software for data processing. All statistical calculations were done through Microsoft Excel 2013.

**Drug Products**

Terbinafine 1% cutaneous solution (Lamisil Once, batch no. 8U6G, Galaxo SmithKline, Brentford, UK) was used as reference formulation and terbinafine 1% FFS (batch no. SMV(7151)080, Sun Pharmaceutical Industries Ltd., Gurugram, India) was used as test formulation. Additionally, for IVRT selectivity, specificity, and sensitivity experiments, two more test formulations were used, i.e. terbinafine 0.5% and 2.0% FFS (batch nos. SMV(7151)088, SMV(7151)090, Sun Pharmaceutical Industries Ltd).

**HPLC-UV Method Validation**

RP-HPLC method validation was performed in the range of 0.501 to 121.000 µg/mL at 283 nm using Zorbax Eclipse
XDB-C8 (150 x 4.6, 5 µ) column from Waters, Hyderabad, India, with the mobile phase comprising of solution 1 (acetonitrile:methanol; 50:50 v/v) in combination with solution 2 (0.3% trifluoroacetic acid in water, v/v) in the ratio of 45:55 v/v, with a 0.700 mL/min flow rate. The injection volume was kept as 10 µL and the column oven temperature was set at 45 °C. We prepared separate stock solutions of TBF in methanol to generate a calibration curve (1 mg/mL) and for preparing quality control samples. The stock solution was further diluted with mobile phase to yield concentrations of 0.501, 1.210, 6.050, 12.100, 24.200, 48.400, 96.800, and 121.000 µg/mL in each analytical run of IVRT samples, a set of eight calibration standards along with one blank were injected. The regression curve was established from all calibration standards. Additionally, at least three levels of QC samples (in the concentrations of 1.253, 48.200, 96.400 µg/mL, i.e., low, middle, and high quality control respectively) were interspersed with each IVRT run.

**IVRT Method**

IVRT was performed using an FDC system, having receptor chamber volume with the capacity of 20 mL. All components of the FDC system such as donor chamber, receptor chamber, clamp, magnetic stirrer, and synthetic membranes were arranged. A magnetic stirrer was placed inside the receptor chamber of FDC. The receptor chamber was filled with receptor solution, i.e., 10× phosphate buffered saline: water, 10:90 v/v containing 7% Ascorbic acid w/v. Membrane was mounted over the receptor chamber until it touched the joint surface between the receptor chamber and the donor chamber. The donor chamber was placed on top of the membrane properly aligning with FDC assembly. A clamp was affixed to stabilize and secure the joint between the donor chamber and the receptor chamber. The underside of each membrane was checked for air bubbles, if any, then it was tilted to remove the air bubbles. The FDC assembly was placed within a cell holder and a port of the water jacket was connected to the tubing of the re-circulator rack. The heating circulator bath was turned on, and the temperature of the water tank was set appropriately to maintain membrane at about 32 ± 1 °C. Magnetic stirrer was turned on (with a fixed speed of approximately 560 rpm) throughout the test. The membrane was kept in equilibrium for 30 min. The temperature on the membrane surface was measured using a calibrated infrared thermometer. Prior to application of formulation on the membrane surface, pre-dose samples (300 µL) were collected from the centre of the receptor chamber from each FDC and transferred into sample collection vials. Stock receptor solution was used for replenishing the receptor chamber after each sampling. The components of the FDC are shown in Figure 1.

**IVRT Apparatus and Laboratory Qualification**

The IVRT apparatus was qualified with all critical parameters of the FDC. These parameters include assessment of the capacity of the receptor chamber, diameter of FDC, temperature on membrane surface and in receptor solution, the stirring speed, dispensing volume, and environment conditions (7). All these parameters are measured by length, weight, and temperature measurement. Results are shown in Table 1.

The laboratory qualification was performed by assessing the release rates of TBF reference formulations with the developed and validated IVRT and HPLC-UV method. Release rates of reference formulation were evaluated on
two different days (n = 6 FDCs per day). Reproducibility, intra- and inter-run variability was determined in term of percent coefficient of variation (%CV). The %CV should be < 15%. The intra-run %CV for the first and second IVRT runs were calculated as 8.98% and 4.01%, respectively; inter-run %CV (n = 12 FDCs) was calculated as 3.82%. The product sameness test was based on computation of 90% confidence interval (CI) as per scale-up and post-approval changes (SUPAC SS) guidance (6). Individual test/reference (T/R) ratios were calculated in percentage terms by considering the first day run as reference and the second day run as test. The 90% CI was subsequently determined from the ordered T/R ratios, where the 8th and 29th ordered individual T/R ratios are the lower and the upper limits, respectively (6). For equivalence, the calculated 90% CI should be within the range of 75–133%. The lower limit and upper limit of 90% CI was calculated as 99.09% and 120.08%, respectively, and the same was falling in the acceptable range.

Table 1. Results of Apparatus Qualification Test

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Acceptance Criteria</th>
<th>Result</th>
<th>Acceptable</th>
</tr>
</thead>
<tbody>
<tr>
<td>FDC capacity (mL)</td>
<td>20 ± 1.0</td>
<td>20 ± 0.1</td>
<td>YES</td>
</tr>
<tr>
<td>Orifice diameter (mm)</td>
<td>15 ± 0.75</td>
<td>15 ± 0.2</td>
<td>YES</td>
</tr>
<tr>
<td>Temperature of receptor solution (°C)</td>
<td>32 ± 1</td>
<td>32 ± 0.4</td>
<td>YES</td>
</tr>
<tr>
<td>Temperature on membrane surface (°C)</td>
<td>32 ± 1</td>
<td>32 ± 0.5</td>
<td>YES</td>
</tr>
<tr>
<td>Speed of magnetic stirrer (rpm)</td>
<td>600 ± 60</td>
<td>565 ± 5</td>
<td>YES</td>
</tr>
<tr>
<td>Dispensed sampling volume (µL)</td>
<td>300 ± 9</td>
<td>302 ± 6</td>
<td>YES</td>
</tr>
</tbody>
</table>

Values are presented as mean ± SD (n = 6). FDC: Franz diffusion cell; SD: standard deviation.

Selection of Receptor Solution
The receptor solution should be selected to maintain the sink conditions throughout the IVRT experiment. In this method, receptor solution was selected based on solubility, reproducibility of results, and R² (which needs to be ≥ 0.90). Solubility of TBF in receptor solution was verified by dissolving 1 mg of TBF in 1 mL of receptor solution.

Selection of Membrane Using Membrane Binding Test Method
This test was performed to determine the suitability of a synthetic membrane in providing inertness for the diffusion of drug across it from a semisolid dosage form. Supor-200, Ultipor, and Tuffryn HT-200 synthetic membranes were used to evaluate membrane binding test. These membranes were dipped in a known concentration of TBF prepared in a receptor solution for at least 6 hours. After the storage period, membranes dipped in TBF solutions were injected and compared with the peak area response got from controlled stock solution.

Formulation Application and Sample Collection
Formulation was applied on the membrane surface of each FDC using a syringe containing approximately 800 µL and was spread evenly. The donor chamber was occluded. At pre-set sampling time points (predose, 0.5, 1, 2, 3, 4, 5, and 6 hours) 0.300 mL of receptor solution was removed from the center of the receptor chamber and transferred into HPLC glass vials. The receptor chamber was replenished with pre-warmed receptor solution.

Selection of Sampling Time Points
Sampling time points were as follows: predose, 0.5, 1, 2, 3, 4, 5, and 6 hours. The sampling time of the experiment for 6 hours was found sufficient to discriminate between different strengths (i.e., 0.5%, 1.0%, and 2.0% w/w) of TBF film forming solution.

IVRT Sample Quantification
After the completion of the experiment, the concentration at each sampling time point was obtained through HPLC analysis. The amount released at each sampling time point (µg/cm²) was calculated as the cumulative concentration obtained at each sampling time point (µg/mL) × volume of FDC (mL) (20 mL) × amount of sample removed from FDC at each sampling time (mL), all divided by effective surface area of membrane (surface area of orifice = 1.77 cm²). Amount of sample removed from FDC in each sampling time (mL) was summed up for the next sample to calculate the cumulative amount removed in previous sampling.

The in vitro release rate calculated by plotting cumulative amount of drug release per unit area (µg/cm²) against time (h¹/²) yields a straight line, the slope of which gives release rate. The percent cumulative drug release was also calculated by dividing the cumulative amount release (µg) by the amount of formulation applied.

Statistical Comparison Between Reference and Test Formulations
Comparison of in vitro release rates was carried out as per the mentioned method in SUPAC-SS guideline. Six slopes (release rates) were obtained from each test and reference formulations. From these slopes, a total of 36 individual T/R ratios (slope of T divided by R) were calculated in terms of percentages. These computed T/R ratios were ranked from lowest to highest. The 8th and 29th ranked individual ratios are the lower and upper limit, respectively, of the 90% CI for the ratio of the in
vitro release rate (slope) for T over R. The 90% CI should fall between 75% to 133% (6).

RESULTS AND DISCUSSION

HPLC Method Validation

Linearity and Range

The linearity of the method was determined by analysis of standard plots associated with an eight-point standard calibration curve. The calibration curve was linear \( y = 21600x + 314 \) from 0.501 µg/mL [limit of quantitation (LOQ)] to 121.000 µg/mL [upper limit of quantitation (ULOQ)] with correlation coefficient \( r = 1.000 \). Back-calculations were performed from that curve to determine concentrations of TBF in unknown samples.

Selectivity Using Synthetic Membrane and Placebo

For the HPLC method validation, the synthetic membrane (Ultipor N66) was dipped in the receptor solution for 6 hours. Similarly, 800 µL of placebo was added to 20 mL of receptor solution, vortexed and kept in bench for 6 hours (covering the entire duration of the experiment). The same was done in triplicate. After analyzing the selectivity samples, the peak area response obtained at RT of analyte was evaluated. As a result, no significant interference was observed at RT of the analyte in selectivity blank samples. This indicates that the analytical method is selective for analysis of TBF from its film forming solution.

Precision and Accuracy

Precision of method is degree of reproducibility and accuracy is degree of exactness. Precision is denoted as % coefficient of variation, and accuracy is denoted as % deviation at each concentration level from nominal concentrations. Here, we have assessed within and between batch precision and accuracy. Within batch precision and accuracy was assessed by analyzing six replicated of quality control (QC) samples, on the same day at three levels of QC (i.e., 1.494, 48.184, and 96.368 µg/mL for low, medium, and high QC, respectively) prepared in receptor solution. The between batch precision and accuracy was assessed by analyzing 18 replicates of QC samples, at each above-mentioned QC level, prepared in receptor solution thorough three precision and accuracy batches ran on two sequential validation days.

The deviation at each QC level from the nominal concentration was found to be 91.00–97.01% for within batch and 95.89–98.16% for between batch accuracy. Similarly, the mean precision was found to be 0.14–1.28% for within batch and 1.03–1.20% for between batch precision.

IVRT METHOD VALIDATION

Solubility of Drug in Receptor Solution

Solubility of TBF in receptor solution, i.e., 10× phosphate buffered saline (water, 10:90 v/v containing 7% ascorbic acid w/v) was found to be excellent. The solubility of 1 mg of TBF in 1 mL of receptor solution was calculated as 762.675 µg/mL (calculated theoretical yield was 763.267 µg/mL), and 99.92% of TBF was calculated as recovery. With this receptor solution, sink condition was maintained, release rates were reproducible and consistent \( R^2 \) was observed throughout the experiments.

Selection of Synthetic Membrane

Among the screened membranes, Supor 200 membrane and Tuffryn HT-200 membrane showed significant binding (7.82% and 8.83% respectively) of TBF on the membrane whereas Ultipor N66 membrane provided the least binding (1.86%). The higher recovery (98.14%) showed that Ultipor N66 membrane provided more inertness in the diffusion of TBF from formulation across it and is therefore more appropriate for IVRT experiments.

Sensitivity, Specificity, and Selectivity

Sensitivity, specificity, and selectivity of this IVRT method was evaluated by investigating the rate of release of TBF from three test FFSs containing 0.5%, 1.0%, and 2.0%. Sensitivity of the IVRT method showed that it can successfully distinguish the three products as the mean release rate increases with increasing the TBF concentration, i.e., 142.22 µg/cm²/h¹/² for 0.5%, 349.53 µg/cm²/h¹/² for 1.0%, and 876.30 µg/cm²/h¹/² for 2.0% (Fig. 2).

A linear regression model was used with the release rate as a dependent variable and TBF concentration of the respective test product as an independent variable to test
for specificity. The results provided evidence of a linear ($R^2 = 0.9478$), proportional relationship between TBF concentration in FFS and the resulting TBF release rates (Fig. 3).

To assess selectivity, the IVRT method should be able to accurately identify in-equivalent product performance. The in-equivalence of the product performance was tested by pair-wise comparisons of the 1.0% TBF FFS with either of the 0.5% or 2.0% TBF FFSs. The pre-determined criterion for equivalence is the calculated 90% CI, i.e., 8th and 29th rank order should be within the range of 75–133%. Here, the computed 8th and 29th rank order is not within the acceptable range (Tables 2 and 3); hence, the in-equivalence of these products was confirmed.

Recovery
Recoveries were evaluated in three runs of IVRT. For each run, six FDCs were used, and reference formulation was applied. Recoveries were calculated for three runs and found to be 21.13%, 20.70%, and 21.45%. The calculated recoveries were less than 30.00%, with an acceptable linearity of TBF release rates throughout the duration; hence, the extent of TBF dose depletion was considered acceptable.

Comparison of Release Rates: Cutaneous Solution Versus Film-Forming Solution
Release rates were calculated for both formulations and are shown in Figure 4. $R^2$ was found to be greater than 0.90 which showed that the rate of drug release from the formulation was consistent over the period of time. The %CV for the release rate was also calculated between each cell (within same day) and found to be less than 15%, which shows low intra-cell variation and reproducibility of the method (Table 3). All the above factors indicated that this developed and validated IVRT method follows Higuchi theory (2, 4).

Table 2. T/R Ratio of Release Rates for Test (0.5%) vs. Reference Formulation (1.0%)

<table>
<thead>
<tr>
<th>Test Release Rate (Slope)</th>
<th>Reference Release Rate (Slope)</th>
<th>T/R Ratios (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>415.3325</td>
<td>368.6100</td>
</tr>
<tr>
<td></td>
<td>427.7148</td>
<td>399.5709</td>
</tr>
<tr>
<td></td>
<td>390.1243</td>
<td>389.8974</td>
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<td>149.1261</td>
<td>278.51</td>
<td>247.18</td>
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<tr>
<td>293.16</td>
<td>390.1243</td>
<td>389.8974</td>
</tr>
<tr>
<td>347.58</td>
<td>286.81*</td>
<td>267.94</td>
</tr>
<tr>
<td>308.48</td>
<td>261.61</td>
<td>261.45</td>
</tr>
<tr>
<td>328.47*</td>
<td>293.16</td>
<td>286.81*</td>
</tr>
<tr>
<td>306.86</td>
<td>340.17</td>
<td>317.78</td>
</tr>
<tr>
<td>299.43</td>
<td>334.39</td>
<td>310.27</td>
</tr>
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<td>299.60</td>
<td>326.48</td>
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<td>326.29</td>
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<td>299.43</td>
<td>296.01</td>
<td>295.84</td>
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</tbody>
</table>

<table>
<thead>
<tr>
<th>Rank Order</th>
<th>8*</th>
<th>4</th>
<th>3</th>
<th>2</th>
</tr>
</thead>
</table>

*indicates 8th rank; #indicates 29th rank.
T: Test (0.5% Terbinafine film-forming solution); R: Reference (1% Terbinafine film-forming solution); CI: confidence interval.

Figure 3. Coefficient of determination ($R^2$) obtained from dissolution profile of terbinafine (0.5%, 1.0%, 2.0%), showing specificity of the method.

Figure 4. Release rates and coefficient of determination ($R^2$) obtained from reference and test formulations containing 1% terbinafine.
Statistical Evaluation via SUPAC-SS
The 90% CI was calculated. As shown in Table 4, the 8th and 29th ranked values are 93.41% and 103.86%, respectively. This indicates the 90% CI is within the limits of 75–133% in accordance with SUPAC-SS guidance.

CONCLUSION
For effective pharmacological action of any topical dosage forms, the formulation must reach the site of action and before that it must be released from the vehicle to penetrate the skin layers. For estimating the release rate from the topical formulations, the IVRT is a useful tool. During the drug development stage, IVRT can be used to finalize the formulation before clinical trials and to check the lot-to-lot consistency. The aim of this study was to establish a fast, accurate, and reproducible in vitro release method for determination of TBF from its topical film-forming solution formulations. In this study, drug release was consistent ($R^2 > 0.99$) and release rates were reproducible (%CV < 15%).
release (recovery) after 6 hours of sampling was less than 30% of the applied formulation, which indicates the sink condition was maintained throughout the experiment. Release rates between test and reference formulations were compared statistically through a widely accepted T/R ratio method, and results fell within 90% CI limits hence, confirming the product sameness.

The developed and validated IVRT and HPLC methods are suitable for drug release testing and routine testing of TBF cutaneous and film-forming solutions. Furthermore, this method can also be applied for determination of release rates of other TBF topical formulations.

**CONFLICT OF INTEREST**
The authors disclosed no conflicts of interest related to this article.

**REFERENCES**


